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can be found if we combine Equations 1.4 and 1.5:  $1.2 \text{ mV} = \frac{2.3 \text{ kT}}{e}$  Assume the temperature is  $22^\circ\text{C}$ . Thermodynamics Solution Manual - CHEME325 - UW - StuDocu Al-Zaytoonah University of Jordan P.O.Box 130 Amman 11733 Jordan Telephone: 00962-6-4291511 00962-6-4291511 Fax: 00962-6-4291432. Email: [president@zuj.edu.jo](mailto:president@zuj.edu.jo). Student Inquiries | [بالتاسف اساتذتنا](mailto:registration@zuj.edu.jo): [registration@zuj.edu.jo](mailto:registration@zuj.edu.jo): [registration@zuj.edu.jo](mailto:registration@zuj.edu.jo) Chemical Engineering Thermodynamics Solution Manual Pdf ... Chemical, Biochemical, and Engineering Thermodynamics Sandler 4th Edition solutions manual \$30.00 solutions manual Thermodynamics with Chemical Engineering

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