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grams of an elements that is numerically equal to the  
atomic mass of the element in amu. Atomic Mass. The  
mass of a mole of any element or compound. Chapter  
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spontaneous, uniform spreading and mixing of  
particles cau.... a process in which gas particles pass  
through a tiny opening i.... the gas law that states that  
the rate of effusion is inversely.... chemistry study  
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g/mol. 8.1.20 1024. Section 10.4 Empirical and  
Molecular Formulas. 1. The percent composition of a  
compound is the percent by mass of each of the  
elements in a compound. 2. Divide the mass of each  
element in the sample by the mass of the sample.  
Then multiply each quotient by 100. 3.c. Ch 10 Study  
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